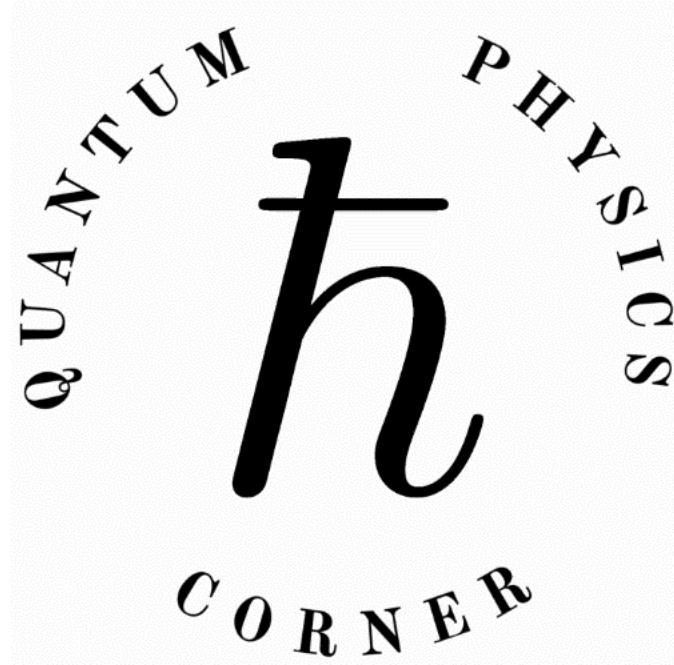




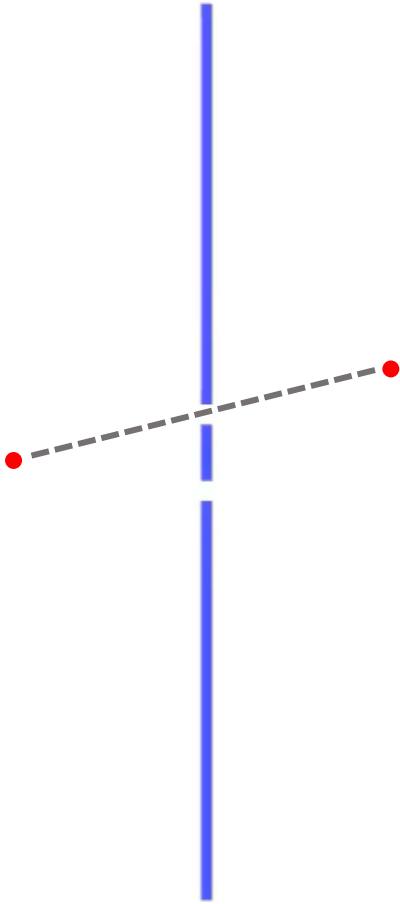
L1 – Quantum Mechanics 1

The Born-Oppenheimer approximation

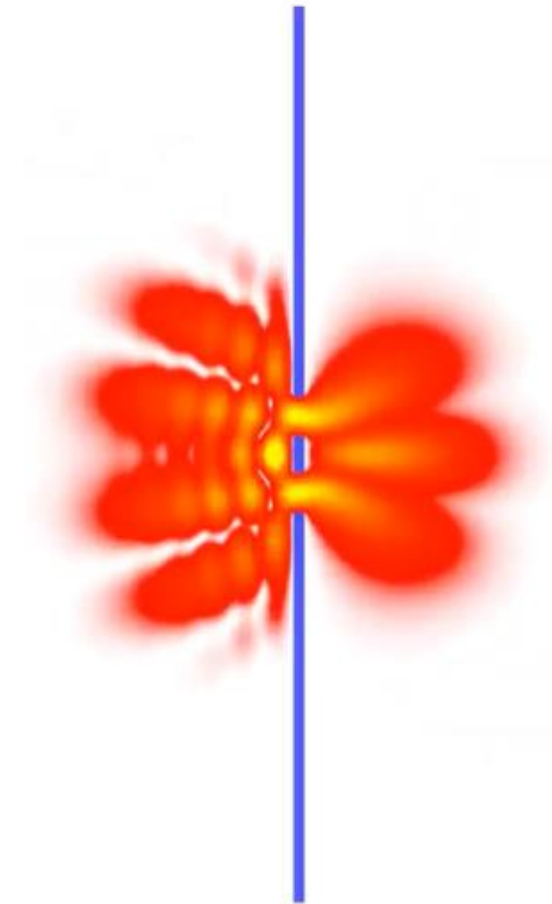
Wave functions and the Schrödinger equation



youtu.be/CnM-pKH2Hi0

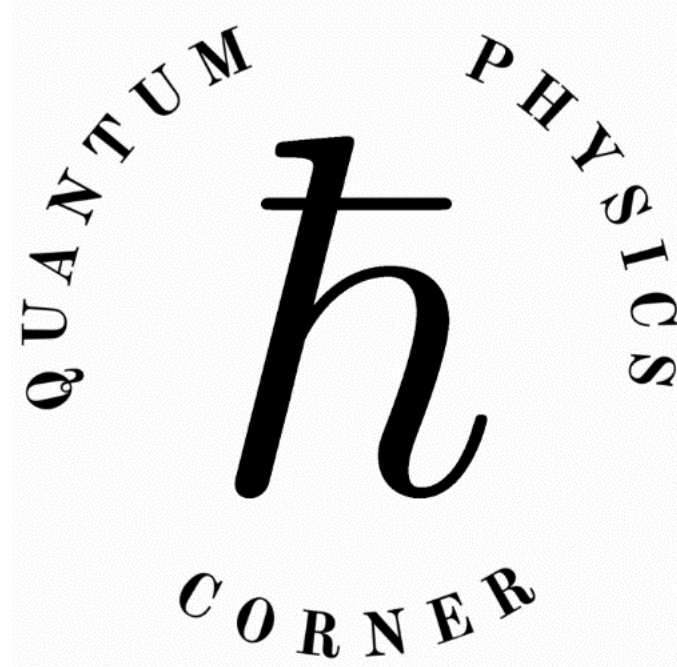


A **classical particle**
is a particle (!)

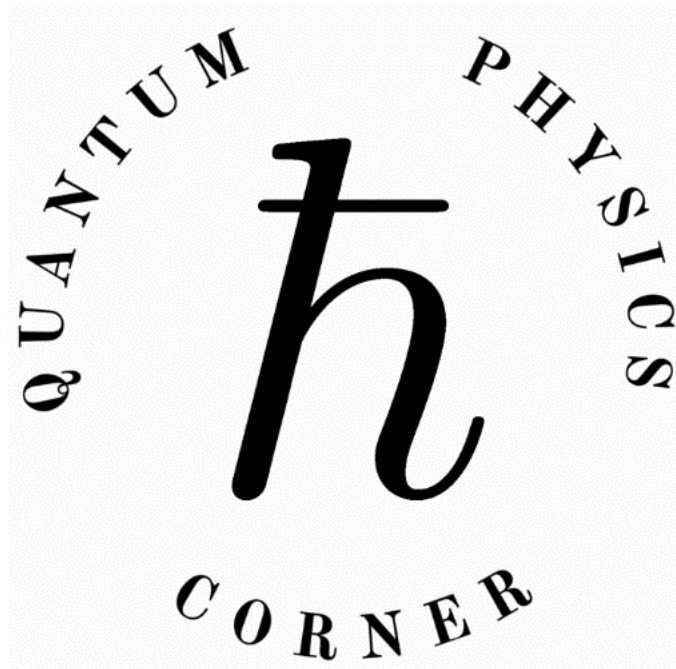


A **quantum particle**
delocalizes over space

The delocalization pattern is determined by the environment.



youtu.be/04zbUVSb6xE



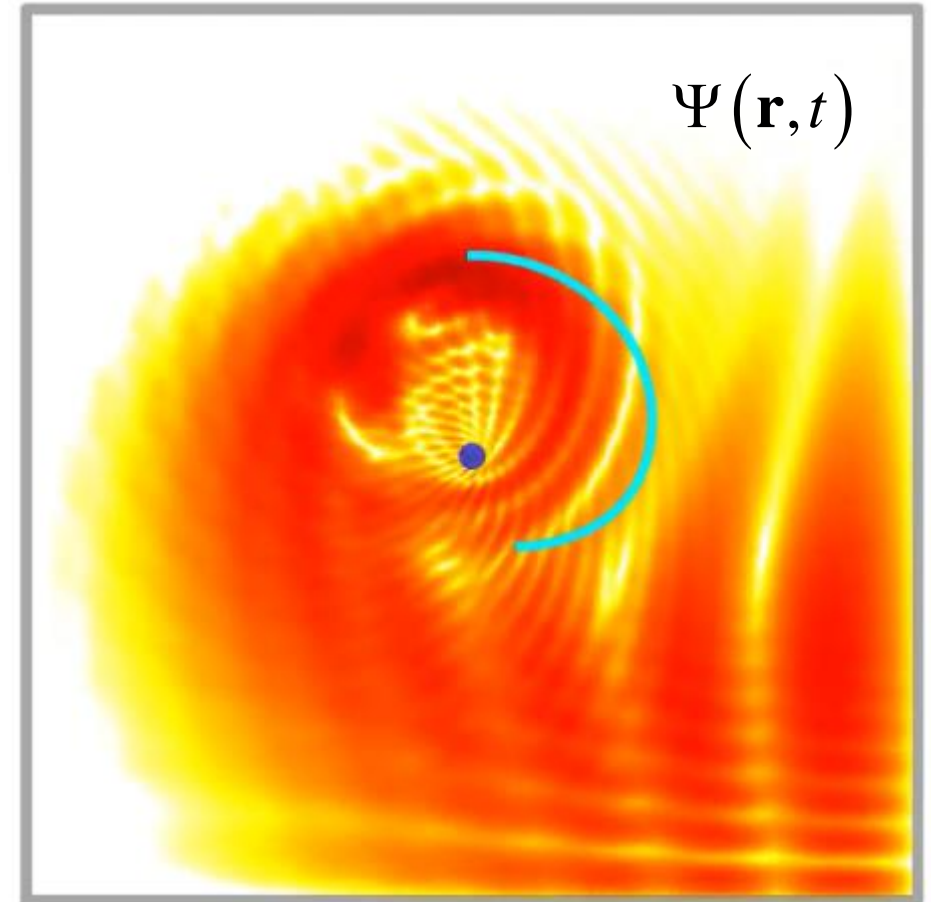
The **quantum delocalization** looks like a wave.

It is mathematically described by a wave function $\Psi(\mathbf{r},t)$.

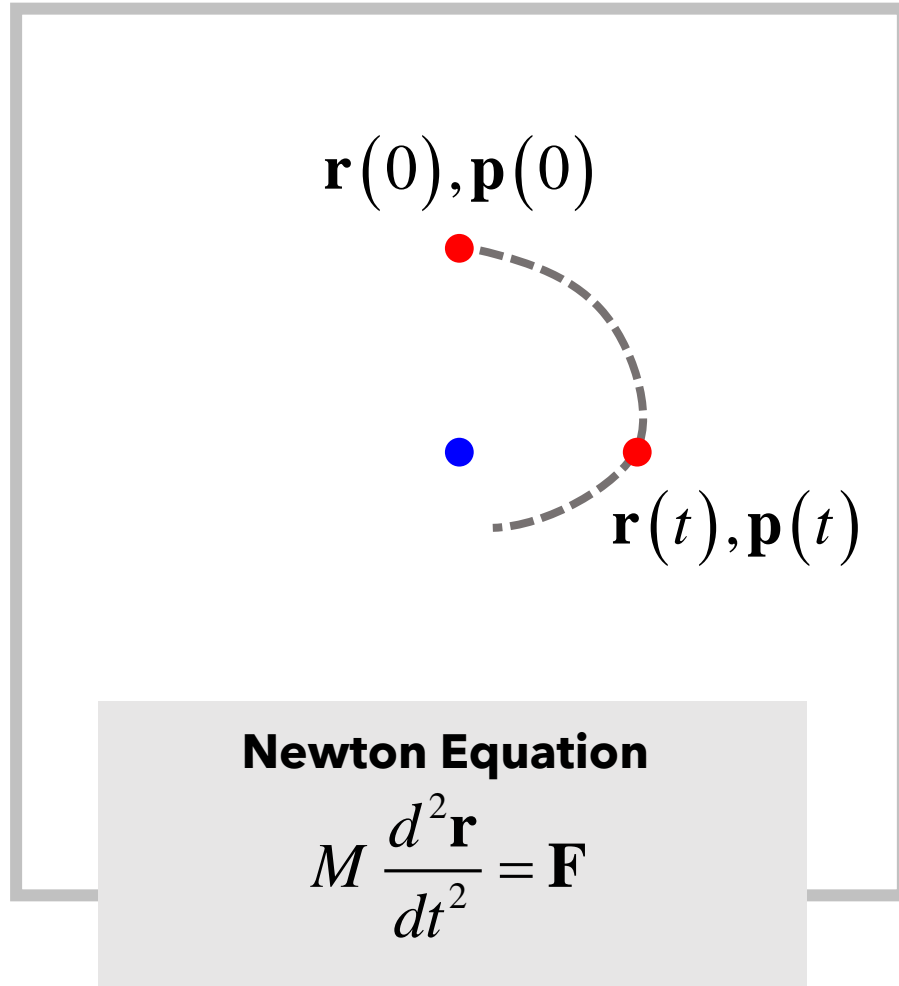
$\Psi(\mathbf{r},t)$ attributes a value for each point \mathbf{r} at each time t .

The value of $\Psi(\mathbf{r},t)$ is a complex number in the form of
 $a + ib$

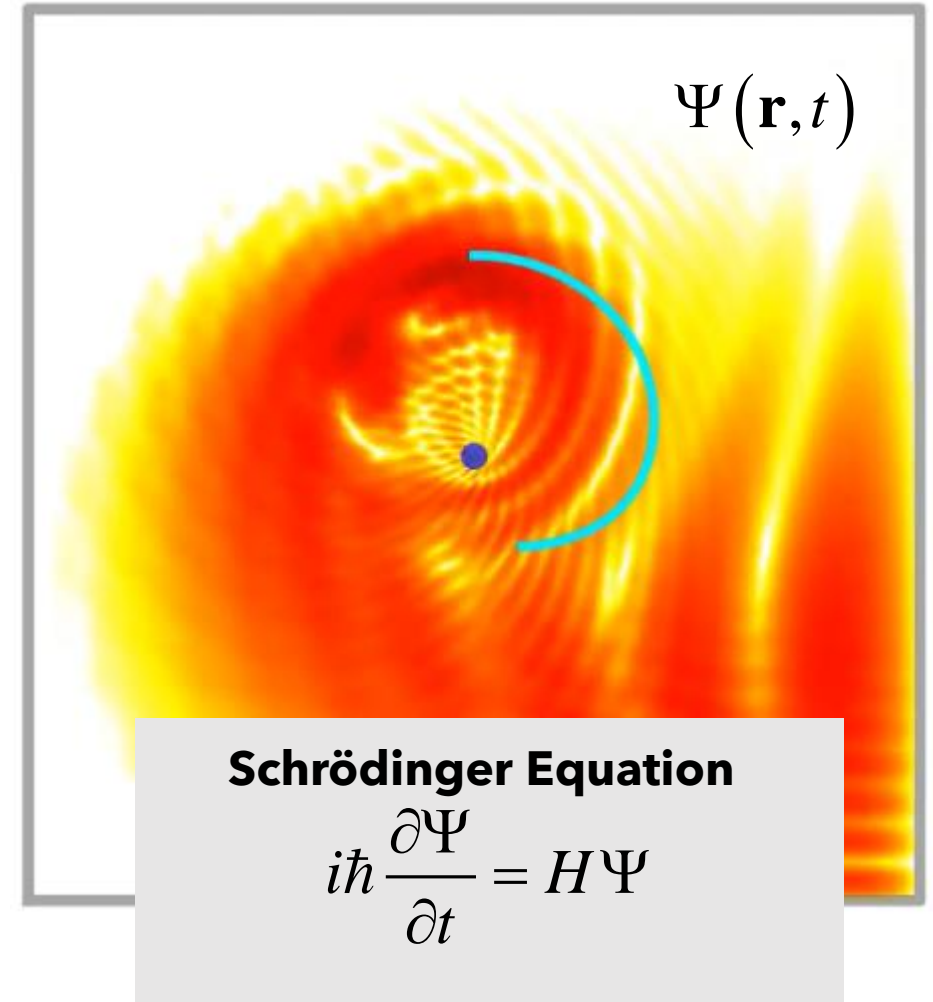
where $i^2 = -1$.



The **classical state** is determined by position and momentum at each time.



The **quantum state** is determined by the wave function of position and time.



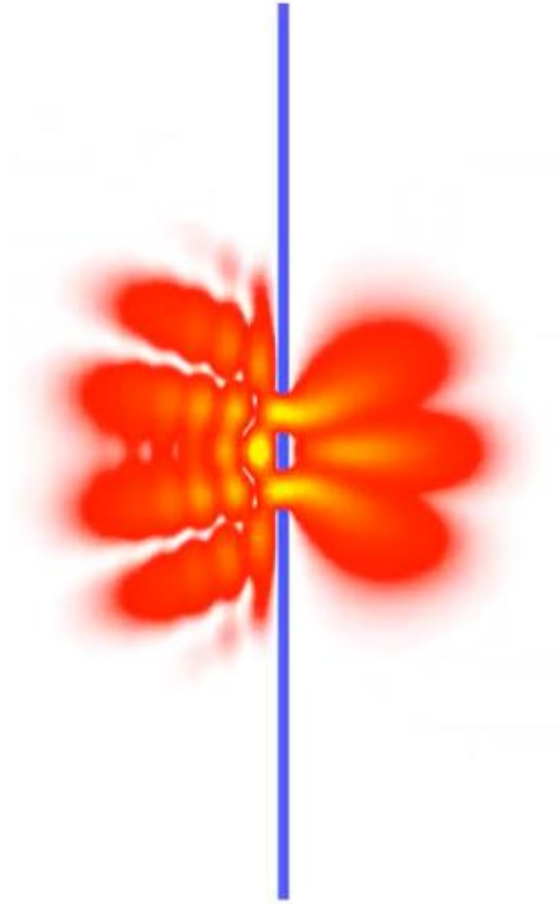
Schrödinger Equation

$$i\hbar \frac{\partial \Psi}{\partial t} = H\Psi$$

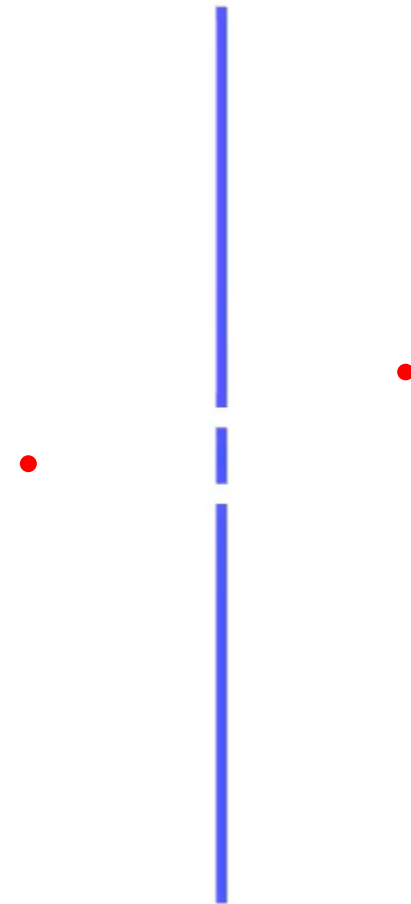
H is the Hamiltonian operator

$\hbar = 6.582 \times 10^{-16}$ eV.s is the Planck constant

We do not directly observe $\Psi(\mathbf{r},t)$.

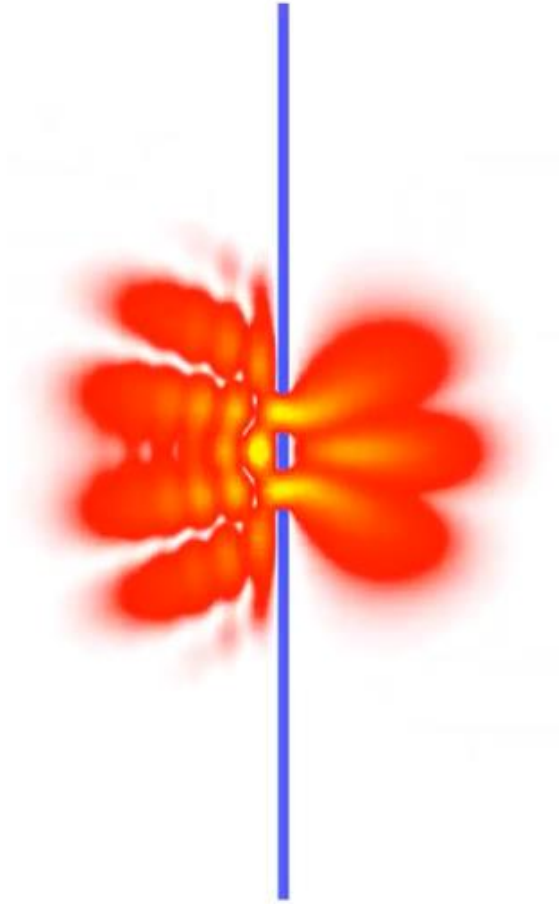


Wave function
 $\Psi(\mathbf{r},t) = a + ib$

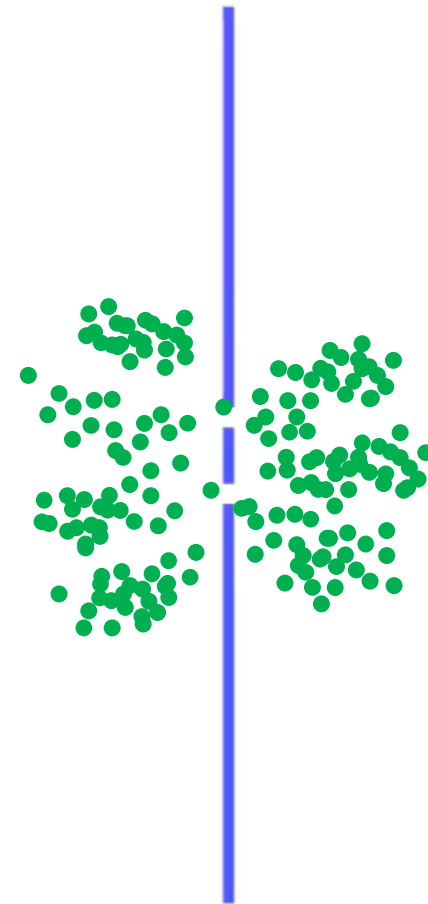


Each measure yields a classical particle.

We do not directly observe $\Psi(\mathbf{r},t)$.



Wave function
 $\Psi(\mathbf{r},t) = a + ib$



Repeated measurements tend to
 $|\Psi(\mathbf{r},t)|^2 = a^2 + b^2$

The Born rule

$$|\Psi(\mathbf{r}, t)|^2 d^3\mathbf{r} = \Psi(\mathbf{r}, t)^* \Psi(\mathbf{r}, t) d^3\mathbf{r}$$

This quantity is the probability of finding the particle in a volume $d^3\mathbf{r}$ around \mathbf{r} at time t .

The probability the particle will be somewhere is 1.

$$\int |\Psi(\mathbf{r}, t)|^2 d^3\mathbf{r} = \int \Psi(\mathbf{r}, t)^* \Psi(\mathbf{r}, t) d^3\mathbf{r} = 1$$

Ψ^* is the complex conjugate of Ψ

$$\Psi = a + ib$$

$$\Psi^* = a - ib$$



WAVE PARTICLE DUALITY

- ▶ wave
- ▶ particle
- ▶ wave packet
- ▶ slit barrier

If an electron approaches through which slit the wave packet, the wave packet will pass only through one of the two slits.



The mean value of any observable A is given as

$$\begin{aligned}\langle A \rangle &= \int d\mathbf{r} \Psi^*(\mathbf{r}, t) A \Psi(\mathbf{r}, t) \\ &= \langle \Psi | A | \Psi \rangle\end{aligned}$$

Bra $\langle \text{bra} |$ and ket $|\text{ket}\rangle$ symbols are Dirac's notation to simplify long integral expressions.

The mean value of the position \mathbf{r}

$$\begin{aligned}\langle \mathbf{r} \rangle &= \int d\mathbf{r} \Psi^*(\mathbf{r}, t) \mathbf{r} \Psi(\mathbf{r}, t) \\ &= \langle \Psi | \mathbf{r} | \Psi \rangle\end{aligned}$$

Mean value of component x

$$\langle x \rangle = \langle \Psi | x | \Psi \rangle = \int dx dy dz \Psi^*(\mathbf{r}, t) x \Psi(\mathbf{r}, t)$$

The mean value of the momentum $p_x = -i\hbar \frac{\partial}{\partial x}$

$$\begin{aligned}\langle p_x \rangle &= -i\hbar \int d\mathbf{r} \Psi^*(\mathbf{r}, t) \frac{\partial \Psi(\mathbf{r}, t)}{\partial x} \\ &= -i\hbar \left\langle \Psi \left| \frac{\partial}{\partial x} \right| \Psi \right\rangle \\ &= -i\hbar \left\langle \Psi \left| \frac{\partial \Psi}{\partial x} \right. \right\rangle\end{aligned}$$

Nabla operator

$$\nabla = \begin{pmatrix} \partial_x \\ \partial_y \\ \partial_z \end{pmatrix} \rightarrow \langle \mathbf{p} \rangle = -i\hbar \langle \Psi | \nabla \Psi \rangle$$

The mean value Hamiltonian H is the energy

$$\begin{aligned}\langle H \rangle &= \int d\mathbf{r} \Psi^*(\mathbf{r}, t) H \Psi(\mathbf{r}, t) \\ &= \langle \Psi | H | \Psi \rangle \\ &= E\end{aligned}$$

Suppose time-independent Hamiltonian $H(\mathbf{r})$.

We can separate \mathbf{r} and t in the wave function:

$$\Psi(\mathbf{r}, t) = \psi(\mathbf{r})\phi(t)$$

Replace in the Schrödinger equation

$$i\hbar \frac{\partial \Psi}{\partial t} = H(\mathbf{r})\Psi \rightarrow i\hbar \frac{\partial \psi(\mathbf{r})\phi(t)}{\partial t} = H(\mathbf{r})\psi(\mathbf{r})\phi(t)$$

$$i\hbar \psi(\mathbf{r}) \frac{d\phi(t)}{dt} = H(\mathbf{r})\psi(\mathbf{r})\phi(t)$$

Separate the variables:

$$i\hbar \frac{1}{\phi(t)} \frac{d\phi(t)}{dt} = \frac{H(\mathbf{r})\psi(\mathbf{r})}{\psi(\mathbf{r})} = E \quad \left\{ \begin{array}{l} i\hbar \frac{d\phi(t)}{dt} = E\phi(t) \\ H(\mathbf{r})\psi(\mathbf{r}) = E\psi(\mathbf{r}) \end{array} \right.$$

The first equation

$$i\hbar \frac{d\phi(t)}{dt} = E\phi(t)$$

gives

$$\phi(t) = \exp\left(-i \frac{Et}{\hbar}\right)$$

The second equation

$$H(\mathbf{r})\psi(\mathbf{r}) = E\psi(\mathbf{r})$$

is called the Time-Independent Schrödinger equation.

Eigenvector



$$H(\mathbf{r})\psi(\mathbf{r}) = E\psi(\mathbf{r})$$



Eigenvalue

$$\begin{array}{l} H(\mathbf{r})\psi_1(\mathbf{r}) = E_1\psi_1(\mathbf{r}) \\ H(\mathbf{r})\psi_2(\mathbf{r}) = E_2\psi_2(\mathbf{r}) \\ \vdots \\ H(\mathbf{r})\psi_N(\mathbf{r}) = E_N\psi_N(\mathbf{r}) \end{array} \left. \begin{array}{l} \text{Ground state} \\ \text{Excited states} \end{array} \right\}$$

Energy



E_4

E_3

E_2

E_1



EXPLANATION

When a quantum particle is confined to a box, it behaves as a wave and has only certain energy levels.



If H is time-independent and $\Psi(r, t=0)$ is an eigenvector of H , the system is in a *stationary state*.

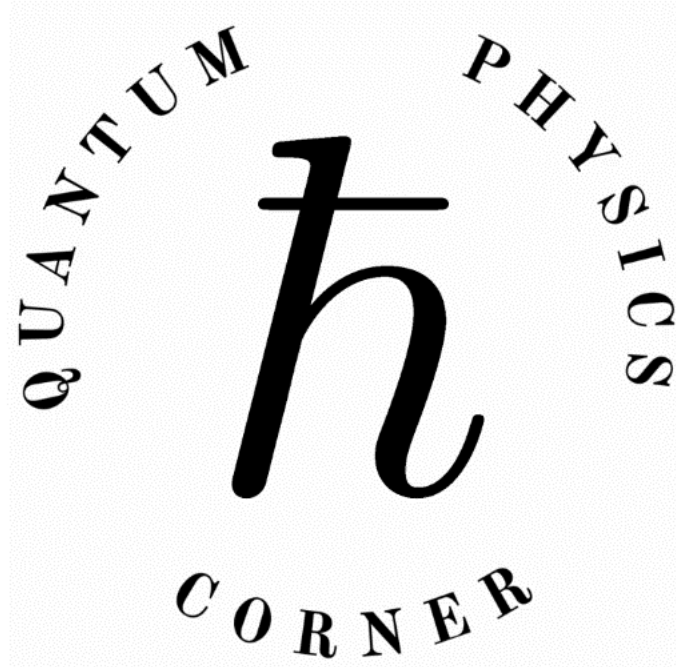
That means that $|\Psi|^2$ remains constant with time.

Oscillations between real and imaginary parts occur with frequency E/\hbar .

If H is time-independent and $\Psi(r, t=0)$ is NOT an eigenvector of H , the system is not *stationary*.

In this case, $|\Psi|^2$ changes with time.

A standard procedure to solve non-stationary states is to first solve a stationary case, and use the eigenvectors to expand the time-dependent Ψ .



If we have more than one particle, we still write a single wave function.

However, the wave function now depends on the coordinates of all particles.

For two particles:

$$\Psi(\mathbf{r}_1, \mathbf{r}_2, t)$$

For N particles:

$$\Psi(\mathbf{r}_1, \mathbf{r}_2, \dots, \mathbf{r}_N, t)$$

The molecular roller coaster

“The Born-Oppenheimer idea is one of those wonderful approximations that even in failure forms the basis for discussion and systematic corrections.

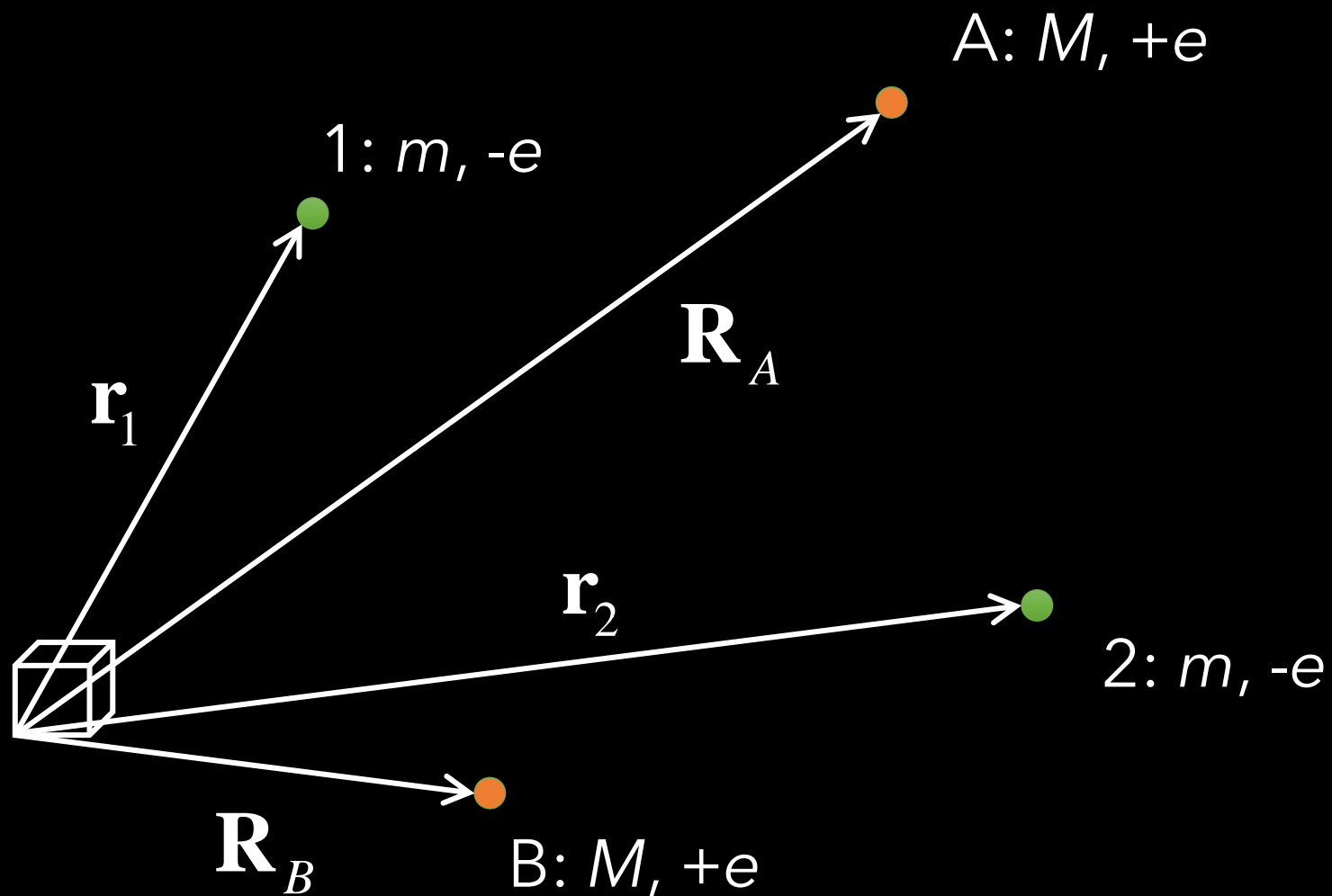
“Without the Born-Oppenheimer approximation as a foundation, there would be no molecular structure, solid-state crystal structure, molecular vibrations, phonons, electronic band structure, and so on.

“Why? Because it is the Born-Oppenheimer approximation that allows separation of electronic from nuclear motion. Without it, we appear to be lost in a soggy many-body ‘pea soup’ or plasma of electrons and nuclei, where there is seemingly no structure at all, save the kind of structure one finds in a two-component liquid.”

Eric J Heller- The semiclassical way to dynamics and spectroscopy, 2018

$$\Psi(\mathbf{R}_A, \mathbf{R}_B, \mathbf{r}_1, \mathbf{r}_2)$$

$$\mathbf{H}\Psi = \varepsilon \Psi$$



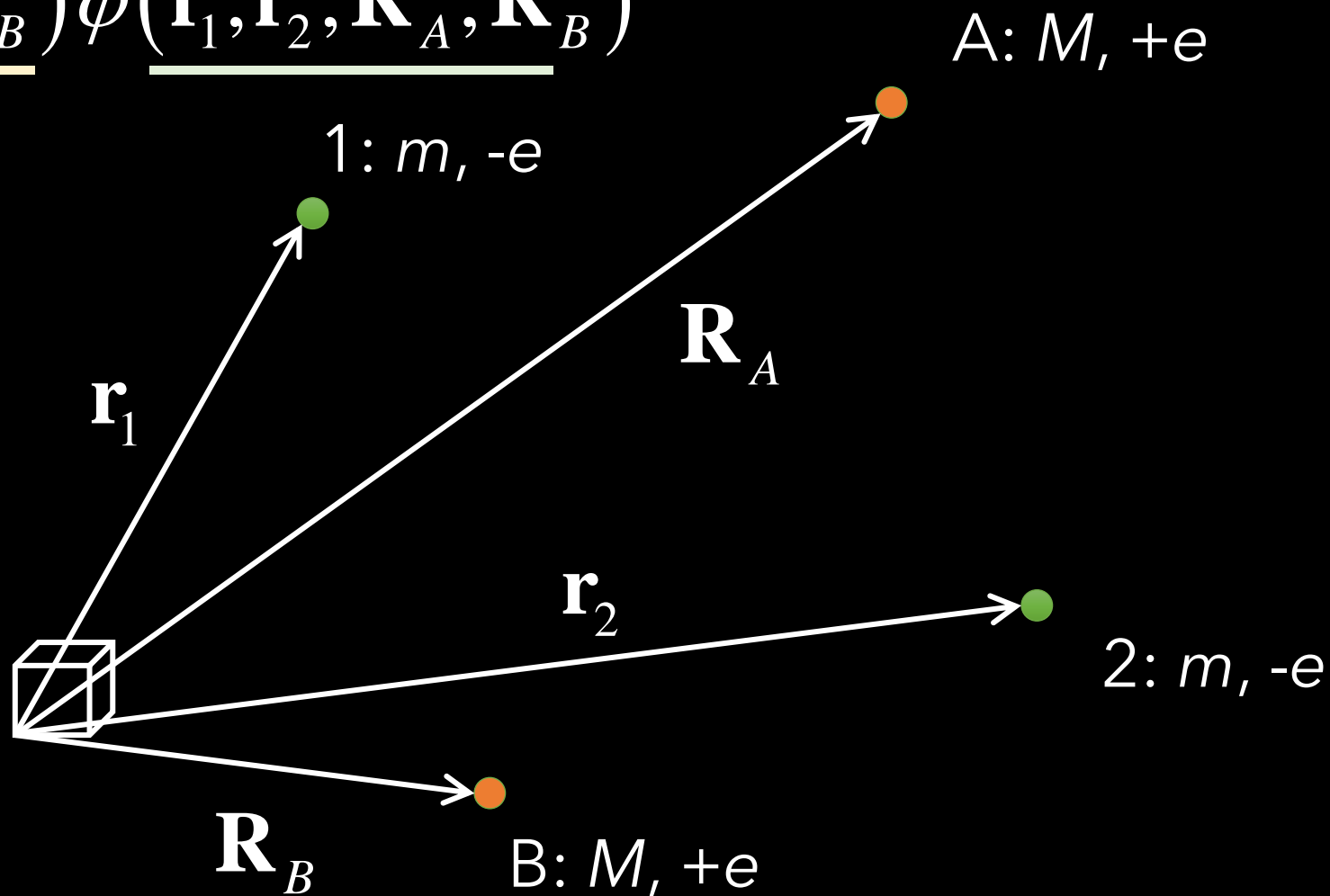
$M = 1836$

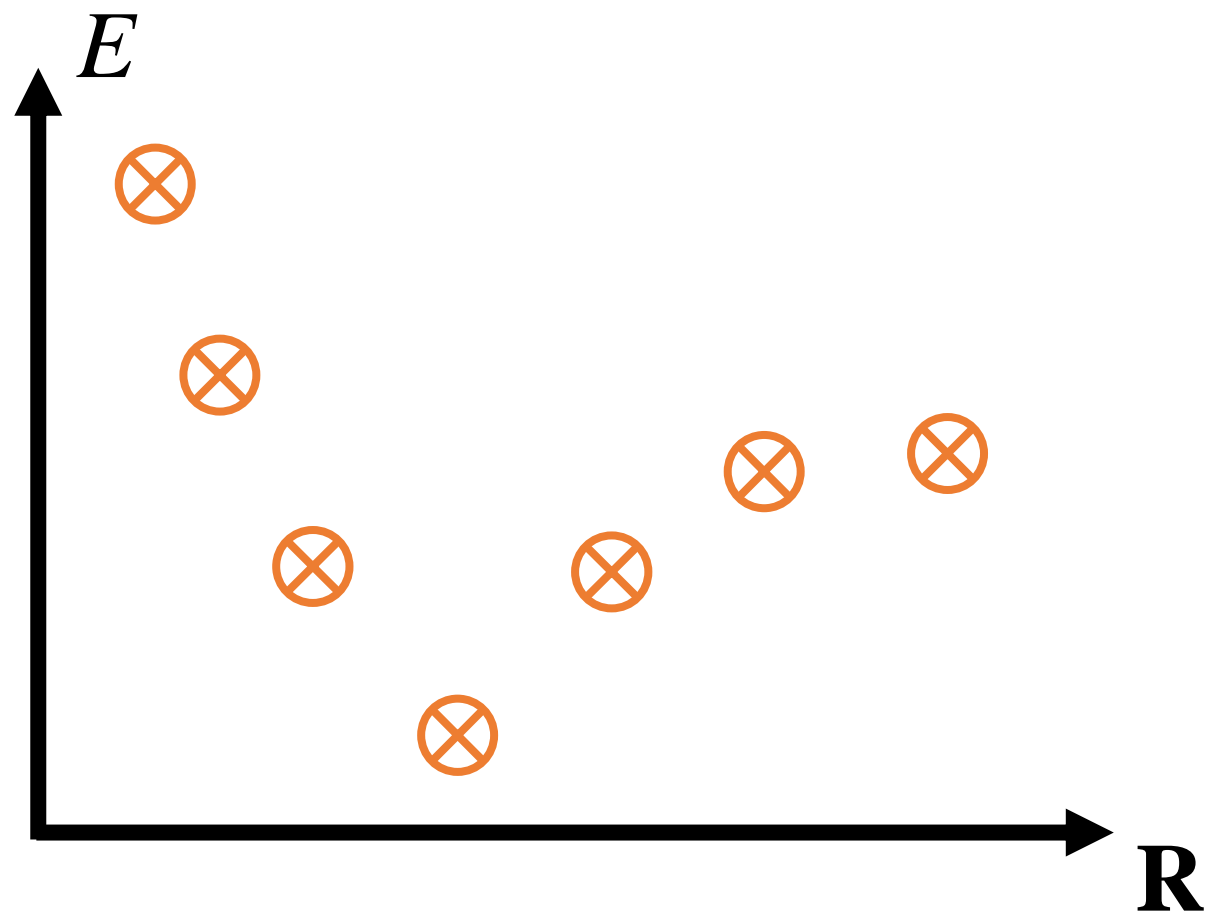
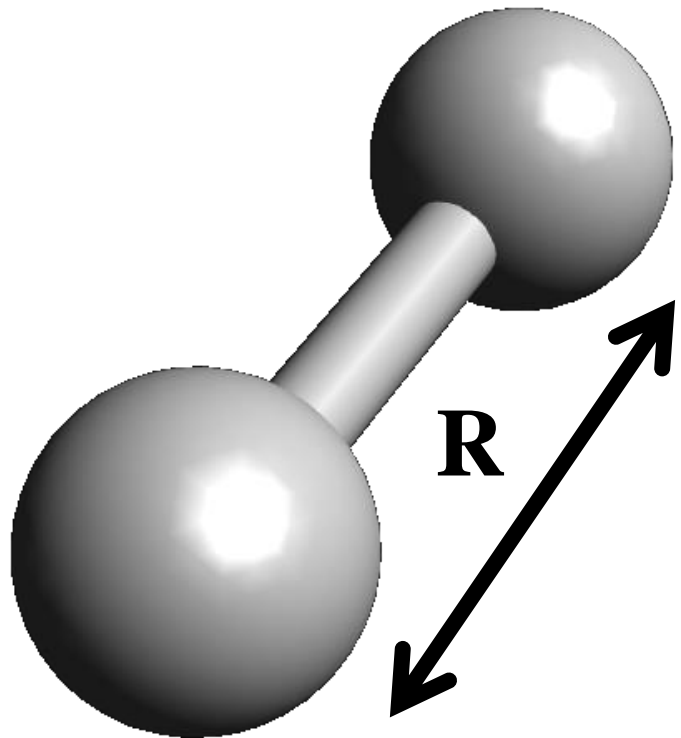


$m = 1$

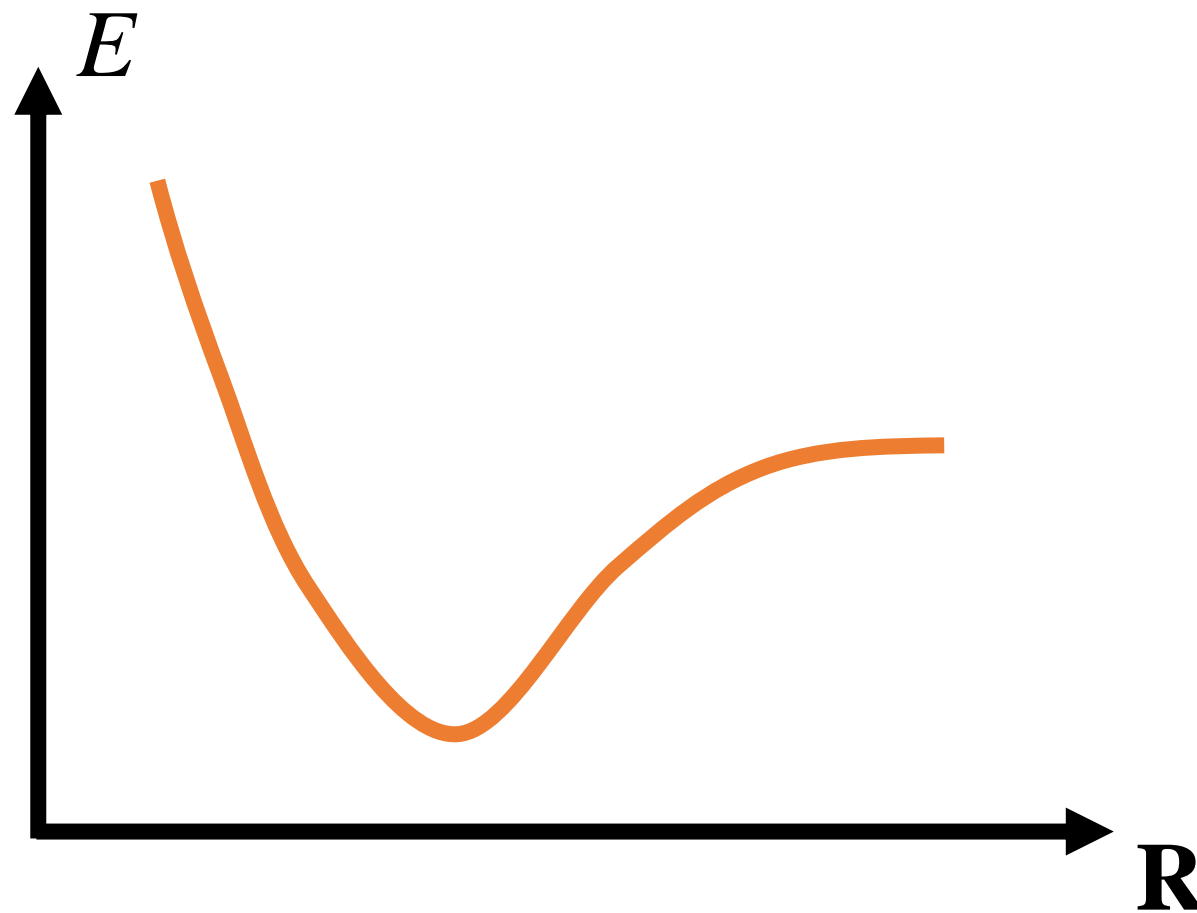
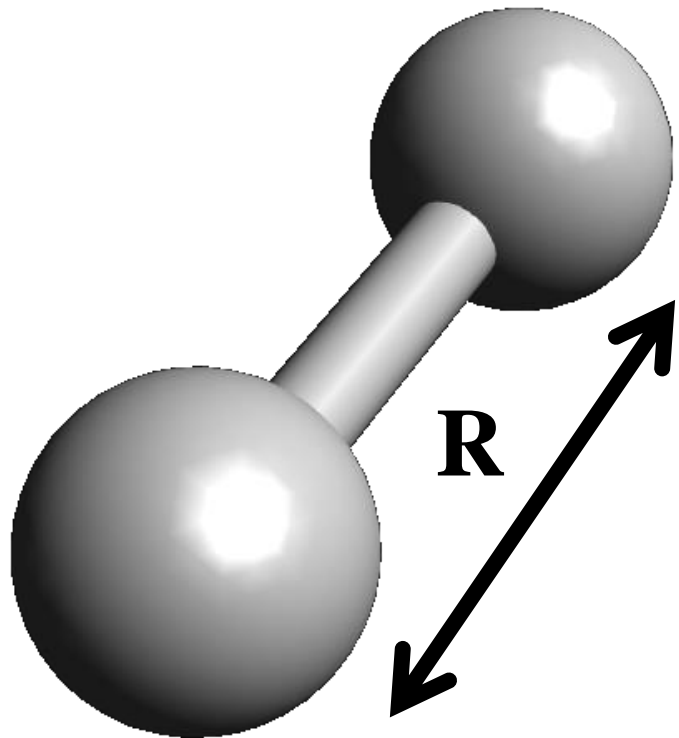
$$\Psi(\mathbf{R}_A, \mathbf{R}_B, \mathbf{r}_1, \mathbf{r}_2)$$

$$= \chi(\mathbf{R}_A, \mathbf{R}_B) \varphi(\mathbf{r}_1, \mathbf{r}_2; \mathbf{R}_A, \mathbf{R}_B)$$



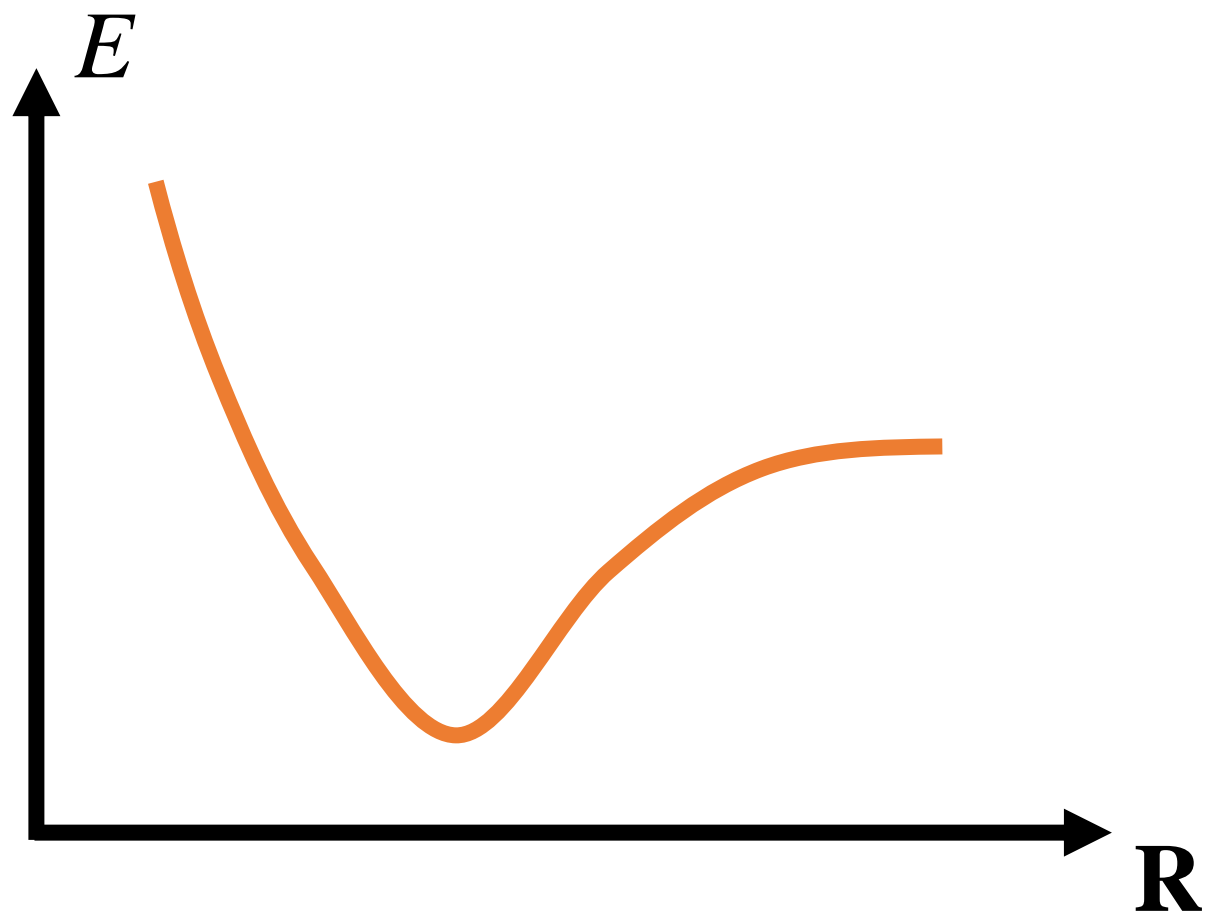


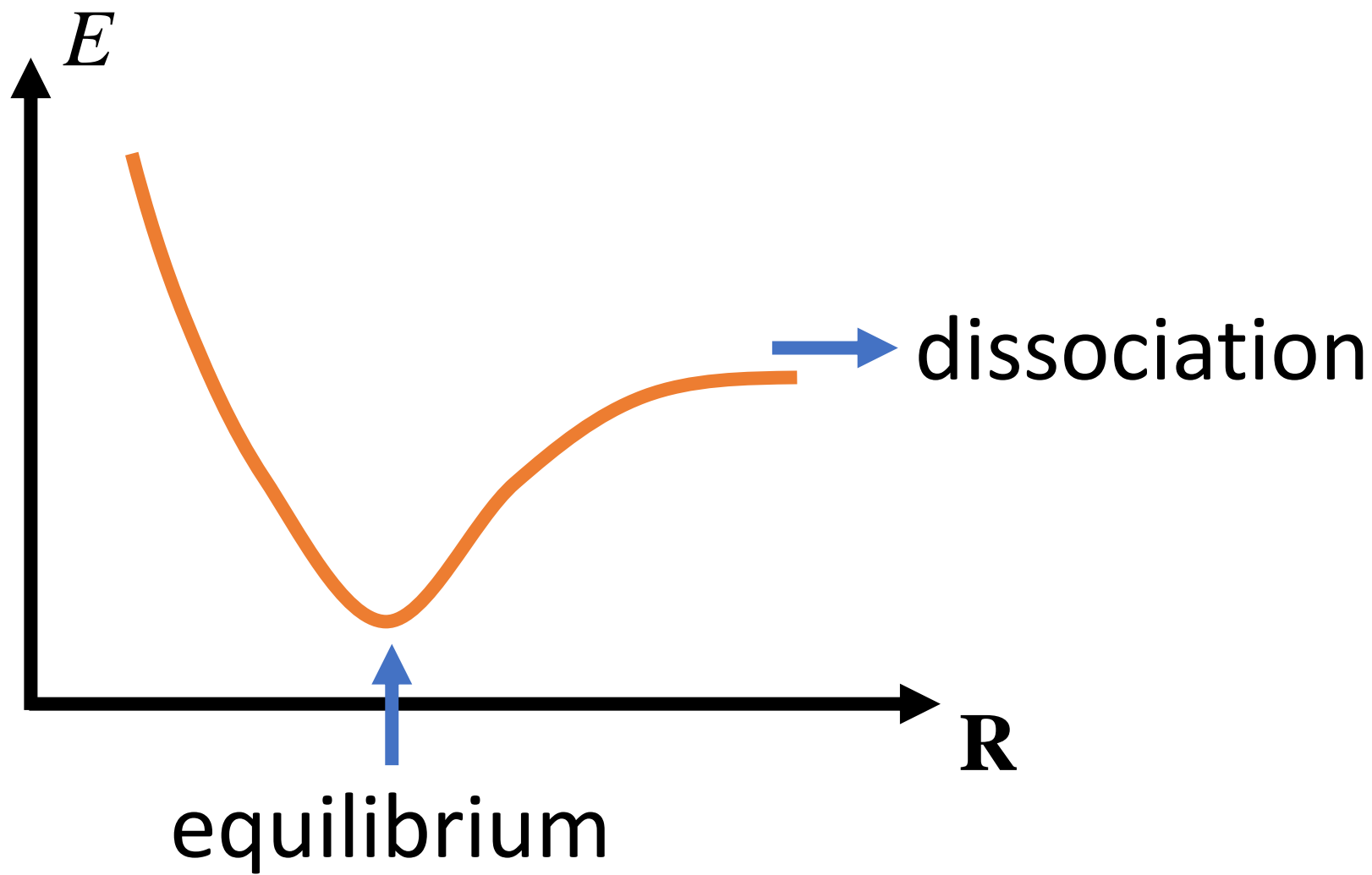
$$H_{elec} \varphi(\mathbf{r}_1, \mathbf{r}_2; \mathbf{R}) = E \varphi(\mathbf{r}_1, \mathbf{r}_2; \mathbf{R})$$

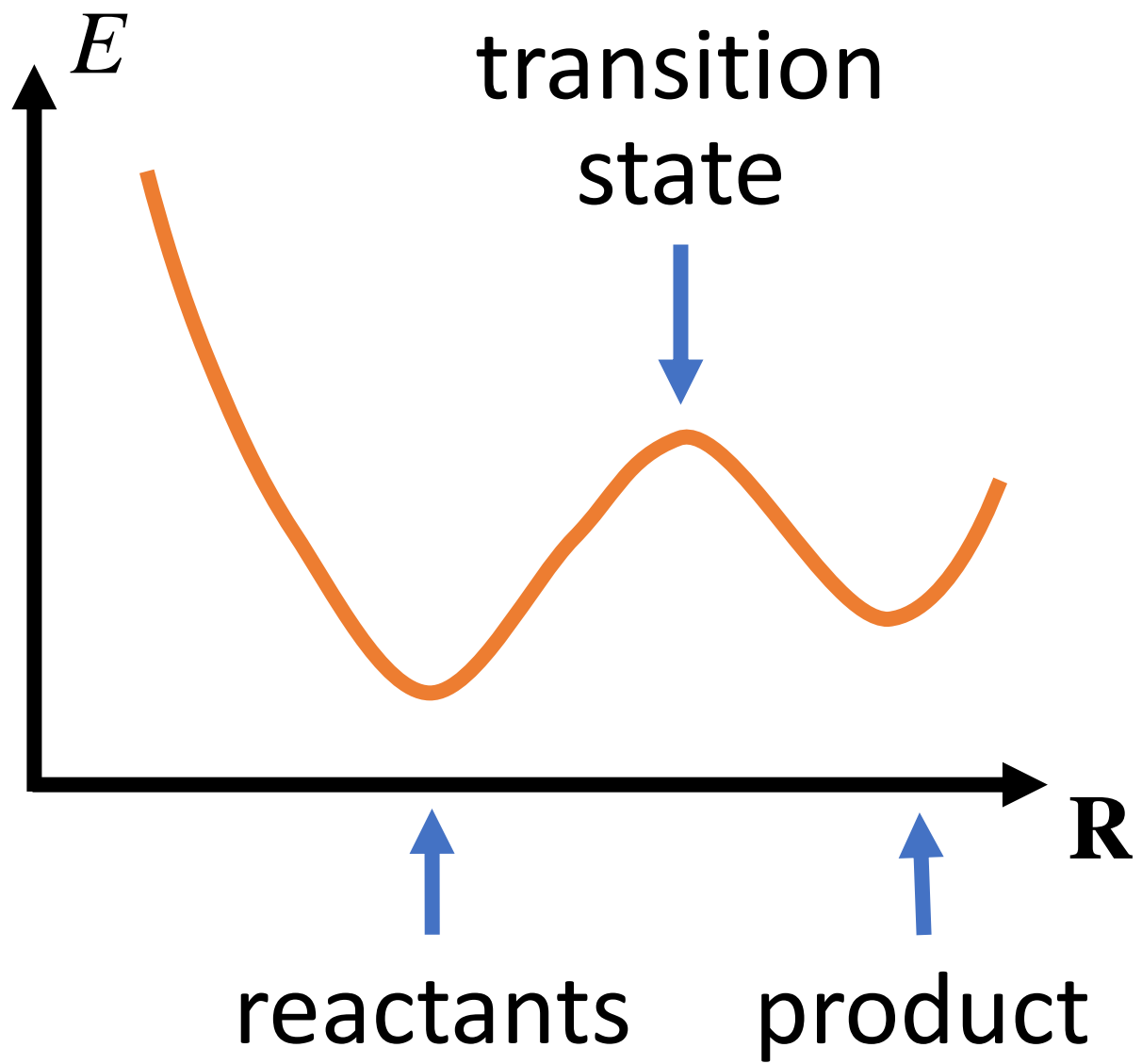


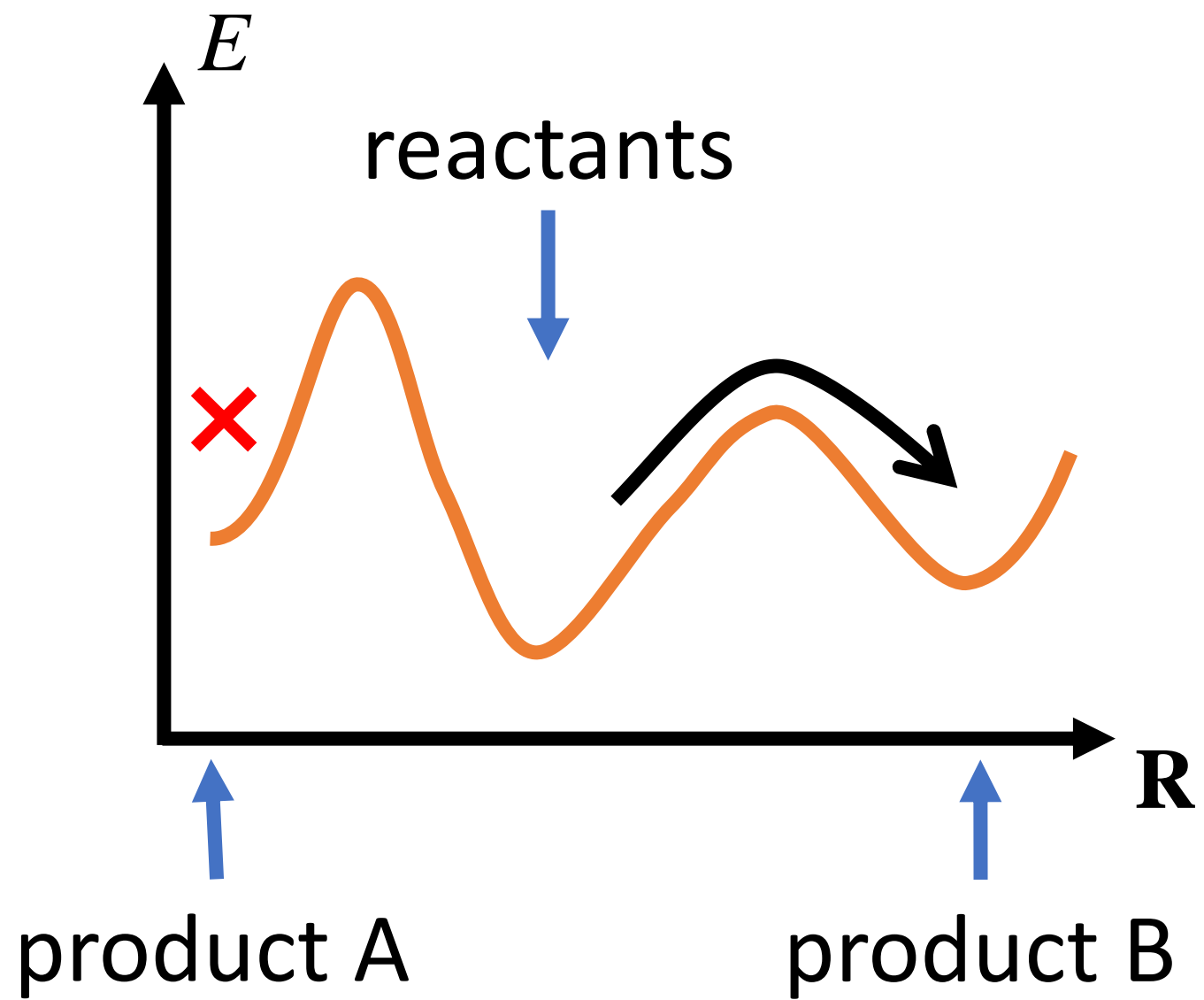
$$H_{elec}\varphi(\mathbf{r}_1, \mathbf{r}_2; \mathbf{R}) = E\varphi(\mathbf{r}_1, \mathbf{r}_2; \mathbf{R})$$

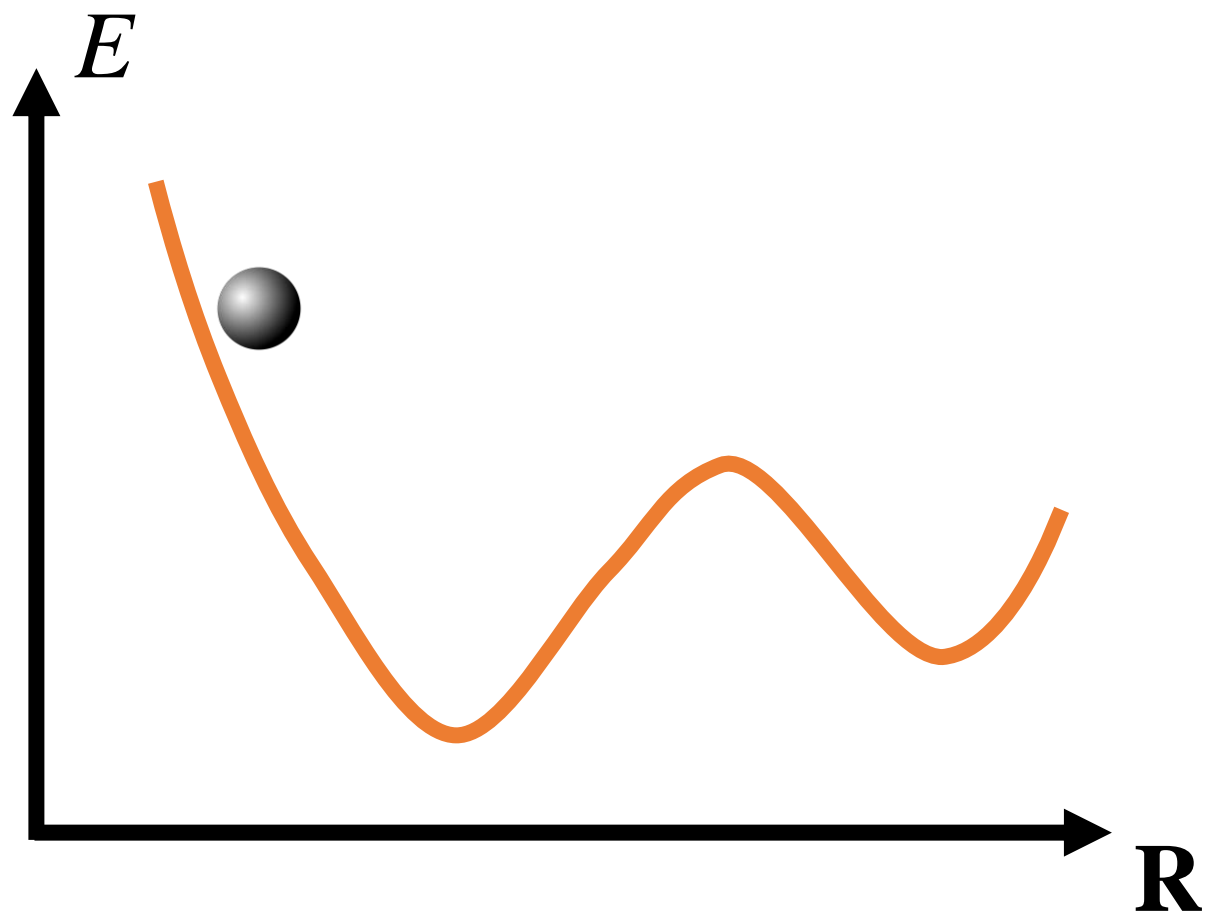
Potential energy surface

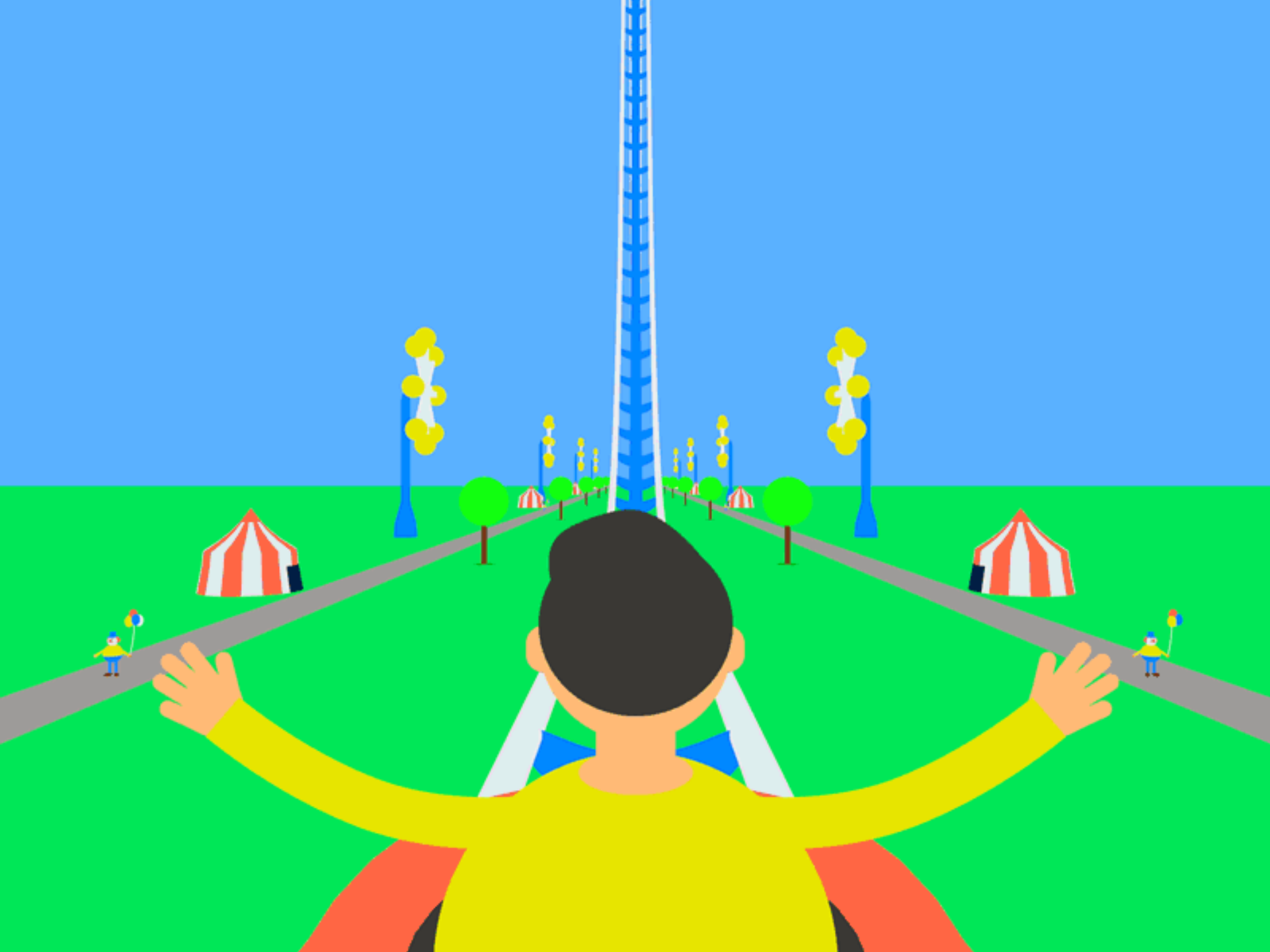


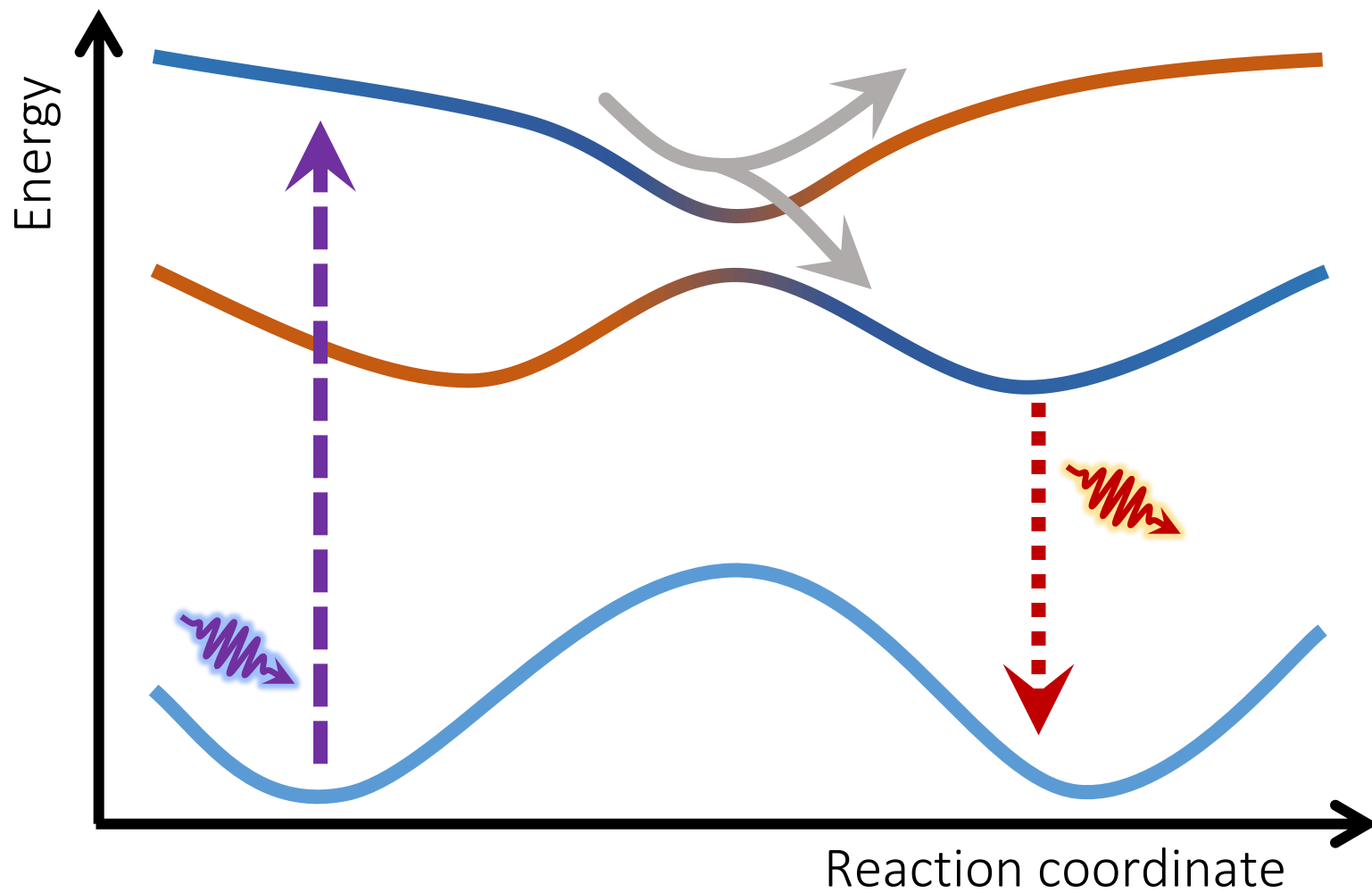












- EM fields
(photoabsorption,
stimulated emission)
- Diabatic mixing
(internal conversion,
intersystem crossing)
- Vacuum fluctuations
(fluorescence,
phosphorescence)

A math reminder

Electronic equation

$$\hat{A}(r, R)\Psi(r, R) = b\Psi(r, R)$$

expand

$$\Psi(r, R) = \sum_n \chi_n(R)\varphi_n(r)$$

replace

$$\hat{A}(r, R)\sum_n \chi_n(R)\varphi_n(r) = b\sum_n \chi_n(R)\varphi_n(r)$$

project

$$\chi_m^*(R)\hat{A}(r, R)\sum_n \chi_n(R)\varphi_n(r) = \chi_m^*(R)b\sum_n \chi_n(R)\varphi_n(r)$$

integrate

$$\int dR \chi_m^*(R)\hat{A}(r, R)\sum_n \chi_n(R)\varphi_n(r) = b\int dR \chi_m^*(R)\sum_n \chi_n(R)\varphi_n(r)$$

group

$$\sum_n \underbrace{\left[\int dR \chi_m^*(R)\hat{A}(r, R)\chi_n(R) \right]}_{C_{mn}(r)} \varphi_n(r) = b \sum_n \underbrace{\left[\int dR \chi_m^*(R)\chi_n(R) \right]}_{D_{mn}} \varphi_n(r)$$

$$\sum_n C_{mn}(r)\varphi_n(r) = b\sum_n D_{mn}\varphi_n(r)$$

Electronic equation

$$\Psi(r, R) \quad \hat{A}(r, R) \Psi(r, R) = b \Psi(r, R)$$

expand

$$\Psi(r, R) = \sum_n \chi_n(R) \varphi_n(r)$$

replace

$$\hat{A}(r, R) \sum_n \chi_n(R) \varphi_n(r) = b \sum_n \chi_n(R) \varphi_n(r)$$

project &
integrate

$$\langle \chi_m(R) | \hat{A}(r, R) \sum_n \chi_n(R) \varphi_n(r) \rangle_R = b \langle \chi_m(R) | \sum_n \chi_n(R) \varphi_n(r) \rangle_R$$

group

$$\sum_n \underbrace{\langle \chi_m(R) | \hat{A}(r, R) | \chi_n(R) \rangle_R}_{C_{mn}(r)} \varphi_n(r) = b \sum_n \underbrace{\langle \chi_m(R) | \chi_n(R) \rangle_R}_{D_{mn}} \varphi_n(r)$$

$$\sum_n C_{mn}(r) \varphi_n(r) = b \sum_n D_{mn} \varphi_n(r)$$

Try yourself to find the electronic equation!

$$\text{For } \hat{A}(r, R)\Psi(r, R) = b\Psi(r, R) \quad (1)$$

$$\Psi(r, R) = \sum_n \chi_n(R)\varphi_n(r) \quad (2)$$

show that

$$\sum_n \langle \chi_m(R) | \hat{A}(r, R) | \chi_n(R) \rangle_R \varphi_n(r) = b \sum_n \langle \chi_m(R) | \chi_n(R) \rangle_R \varphi_n(r)$$

Hint:

- Replace Eq. (2) in (1)
- Project: multiply by $\langle \chi_m(R) |$ at left of both sides

Nuclear equation

$$\hat{A}(r, R)\Psi(r, R) = b\Psi(r, R)$$

expand

$$\Psi(r, R) = \sum_n \chi_n(R) \varphi_n(r)$$

replace

$$\hat{A}(r, R) \sum_n \chi_n(R) \varphi_n(r) = b \sum_n \chi_n(R) \varphi_n(r)$$

project &
integrate

$$\langle \varphi_m(r) | \hat{A}(r, R) \sum_n \chi_n(R) \varphi_n(r) \rangle_r = b \langle \varphi_m(r) | \sum_n \chi_n(R) \varphi_n(r) \rangle_r$$

group

$$\sum_n \underbrace{\langle \varphi_m(r) | \hat{A}(r, R) | \varphi_n(r) \rangle_r}_{E_{mn}(R)} \chi_n(R) = b \sum_n \underbrace{\langle \varphi_m(r) | \varphi_n(r) \rangle_r}_{F_{mn}} \chi_n(R)$$

$$\sum_n E_{mn}(R) \chi_n(R) = b \sum_n F_{mn} \chi_n(R)$$

$$\hat{A}(r, R)\Psi(r, R) = b\Psi(r, R)$$

~~$$\sum_n C_{mn}(r)\varphi_n(r) = b\sum_n D_{mn}\varphi_n(r)$$~~

$$\sum_n E_{mn}(R)\chi_n(R) = b\sum_n F_{mn}\chi_n(R)$$

$$\Psi(r, R) = \sum_n \chi_n(R)\varphi_n(r)$$

$$b = \left\langle \Psi(r, R) \left| \hat{A}(r, R) \right| \Psi(r, R) \right\rangle_{r, R}$$

$$\hat{A}(r, R)\Psi(r, R) = b\Psi(r, R)$$

$$\hat{A}^{(0)}(r, R)\varphi(r; R) = \varepsilon\varphi(r; R)$$

$$\sum_n E_{mn}(R)\chi_n(R) = b\sum_n F_{mn}\chi_n(R)$$

$$\Psi(r, R) = \sum_n \chi_n(R)\varphi_n(r)$$

$$b = \left\langle \Psi(r, R) \left| \hat{A}(r, R) \right| \Psi(r, R) \right\rangle_{r, R}$$

The Born-Oppenheimer approximation

Field-free non-relativistic molecular problem

$$H(\mathbf{R}, \mathbf{r}) \Psi^k(\mathbf{R}, \mathbf{r}) = \varepsilon_k \Psi^k(\mathbf{R}, \mathbf{r})$$

with

$$H(\mathbf{R}, \mathbf{r}) = T_{nuc}(\mathbf{R}) + T_{elec}(\mathbf{r}) + V(\mathbf{r}, \mathbf{R})$$

Born-Huang wave function

$$\Psi_k(\mathbf{R}, \mathbf{r}) = \sum_{nj} c_{nj}^k \varphi_n(\mathbf{r}; \mathbf{R}) \chi_{nj}(\mathbf{R})$$

$$\begin{aligned} \Psi_k(\mathbf{R}, \mathbf{r}) &= \sum_n \varphi_n(\mathbf{r}; \mathbf{R}) \left(\sum_j c_{nj}^k \chi_{nj}(\mathbf{R}) \right) \\ &= \sum_n \varphi_n(\mathbf{r}; \mathbf{R}) h_n^k(\mathbf{R}) \end{aligned}$$

Solving the **electronic** part

$$\Psi_k(\mathbf{R}, \mathbf{r}) = \sum_n \varphi_n(\mathbf{r}; \mathbf{R}) h_n^k(\mathbf{R})$$

$$\left(T_{elec}(\mathbf{r}) + V(\mathbf{r}, \mathbf{R}) \right) \varphi_n(\mathbf{r}; \mathbf{R}) = E_n(\mathbf{R}) \varphi_n(\mathbf{r}; \mathbf{R})$$

Solving the **nuclear** part

$$H(\mathbf{R}, \mathbf{r}) \Psi^k(\mathbf{R}, \mathbf{r}) = \varepsilon_k \Psi^k(\mathbf{R}, \mathbf{r})$$

with

$$H = T_{nuc}(\mathbf{R}) + T_{elec}(\mathbf{r}) + V(\mathbf{r}, \mathbf{R})$$

$$\Psi^k(\mathbf{R}, \mathbf{r}) = \sum_n \varphi_n(\mathbf{r}; \mathbf{R}) h_n^k(\mathbf{R})$$

$$(T_{nuc}(\mathbf{R}) + T_{elec}(\mathbf{r}) + V(\mathbf{r}, \mathbf{R})) \left(\sum_n \varphi_n(\mathbf{r}; \mathbf{R}) h_n^k(\mathbf{R}) \right) = \varepsilon_k \left(\sum_n \varphi_n(\mathbf{r}; \mathbf{R}) h_n^k(\mathbf{R}) \right)$$

$$\left(T_{nuc}(\mathbf{R}) + T_{elec}(\mathbf{r}) + V(\mathbf{r}, \mathbf{R}) \right) \left(\sum_n \varphi_n(\mathbf{r}; \mathbf{R}) h_n^k(\mathbf{R}) \right) = \varepsilon_k \left(\sum_n \varphi_n(\mathbf{r}; \mathbf{R}) h_n^k(\mathbf{R}) \right)$$

Working on the left-side term

$$T_{nuc} \left(\sum_n \varphi_n h_n^k \right) + (T_{elec} + V) \left(\sum_n \varphi_n h_n^k \right) =$$

$$-\frac{\hbar^2}{2\mathbf{M}} \nabla_{\mathbf{R}}^2 \left(\sum_n \varphi_n h_n^k \right) + \sum_n E_n \varphi_n h_n^k$$

$$\frac{1}{\mathbf{M}} f(\mathbf{R}) \rightarrow \sum_{\alpha} \frac{1}{M_{\alpha}} f(\mathbf{R}_{\alpha})$$

$$-\frac{\hbar^2}{2\mathbf{M}} \nabla_{\mathbf{R}}^2 \left(\sum_n \varphi_n(\mathbf{r}; \mathbf{R}) h_n^k(\mathbf{R}) \right) + \sum_n E_n \varphi_n h_n^k = \varepsilon_k \left(\sum_n \varphi_n h_n^k \right)$$

Expanding the blue term

$$-\frac{\hbar^2}{2\mathbf{M}} \nabla_{\mathbf{R}}^2 \left(\sum_n \varphi_n h_n^k \right) =$$

$$-\frac{\hbar^2}{2\mathbf{M}} \sum_n \left[\left(\nabla_{\mathbf{R}}^2 \varphi_n \right) h_n^k + 2 \nabla_{\mathbf{R}} \varphi_n \cdot \nabla_{\mathbf{R}} h_n^k + \varphi_n \nabla_{\mathbf{R}}^2 h_n^k \right]$$

$$\begin{aligned}
& -\frac{\hbar^2}{2\mathbf{M}} \sum_n \left[\left(\nabla_{\mathbf{R}}^2 \varphi_n \right) h_n^k + 2 \nabla_{\mathbf{R}} \varphi_n \cdot \nabla_{\mathbf{R}} h_n^k + \varphi_n \nabla_{\mathbf{R}}^2 h_n^k \right] \\
& + \sum_n E_n \varphi_n h_n^k = \varepsilon_k \left(\sum_n \varphi_n h_n^k \right)
\end{aligned}$$

Projecting on n'

$$\begin{aligned}
& - \left\langle \varphi_{n'} \left| \frac{\hbar^2}{2\mathbf{M}} \sum_n \left[\left(\nabla_{\mathbf{R}}^2 \varphi_n \right) h_n^k + 2 \nabla_{\mathbf{R}} \varphi_n \cdot \nabla_{\mathbf{R}} h_n^k + \varphi_n \nabla_{\mathbf{R}}^2 h_n^k \right] \right\rangle_{\mathbf{r}} \\
& + \left\langle \varphi_{n'} \left| \sum_n E_n \varphi_n h_n^k \right\rangle_{\mathbf{r}} = \left\langle \varphi_{n'} \left| \varepsilon_k \left(\sum_n \varphi_n h_n^k \right) \right\rangle_{\mathbf{r}}
\end{aligned}$$

$$\begin{aligned}
& - \left\langle \varphi_{n'} \left| \frac{\hbar^2}{2M} \sum_n \left[(\nabla_{\mathbf{R}}^2 \varphi_n) h_n^k + 2 \nabla_{\mathbf{R}} \varphi_n \cdot \nabla_{\mathbf{R}} h_n^k + \varphi_n \nabla_{\mathbf{R}}^2 h_n^k \right] \right. \right\rangle_{\mathbf{r}} \\
& + \left\langle \varphi_{n'} \left| \sum_n E_n \varphi_n h_n^k \right. \right\rangle_{\mathbf{r}} = \left\langle \varphi_{n'} \left| \varepsilon_k \left(\sum_n \varphi_n h_n^k \right) \right. \right\rangle_{\mathbf{r}}
\end{aligned}$$

Using orthonormality

$$\begin{aligned}
& \langle \varphi_{n'} | \varphi_n \rangle_{\mathbf{r}} = \delta_{nn'} \\
& - \frac{\hbar^2}{2M} \sum_n \left[\langle \varphi_{n'} | \nabla_{\mathbf{R}}^2 \varphi_n \rangle_{\mathbf{r}} h_n^k + 2 \langle \varphi_{n'} | \nabla_{\mathbf{R}} \varphi_n \rangle_{\mathbf{r}} \cdot \nabla_{\mathbf{R}} h_n^k \right] \\
& - \frac{\hbar^2}{2M} \nabla_{\mathbf{R}}^2 h_{n'}^k + E_{n'} h_{n'}^k = \varepsilon_k h_{n'}^k
\end{aligned}$$

Time-independent Born-Huang formulation

$$H_{n'} h_{n'}^k - \varepsilon_k h_{n'}^k + \sum_n N_{n'n} = 0$$

$$H_{n'} = -\frac{\hbar^2}{2\mathbf{M}} \nabla_{\mathbf{R}}^2 + E_{n'}$$

$$N_{n'n} = -\frac{\hbar^2}{2\mathbf{M}} \left[\langle \varphi_{n'} | \nabla_{\mathbf{R}}^2 \varphi_n \rangle_{\mathbf{r}} h_n^k + 2 \langle \varphi_{n'} | \nabla_{\mathbf{R}} \varphi_n \rangle_{\mathbf{r}} \cdot \nabla_{\mathbf{R}} h_n^k \right]$$

$$\begin{pmatrix} H_1(\mathbf{R}) - \varepsilon_k & N_{12}(\mathbf{R}) & N_{13}(\mathbf{R}) & \dots \\ N_{21}(\mathbf{R}) & H_2(\mathbf{R}) - \varepsilon_k & N_{23}(\mathbf{R}) & \dots \\ \vdots & \vdots & \vdots & \dots \end{pmatrix} \begin{pmatrix} h_1^k(\mathbf{R}) \\ h_2^k(\mathbf{R}) \\ h_3^k(\mathbf{R}) \\ \vdots \end{pmatrix} = \begin{pmatrix} 0 \\ 0 \\ 0 \\ \vdots \end{pmatrix}$$

Adiabatic approximation

$$N_{n'n}(\mathbf{R}) = 0$$

$$\begin{pmatrix} H_1(\mathbf{R}) - \varepsilon_k & 0 & 0 & \dots \\ 0 & H_2(\mathbf{R}) - \varepsilon_k & 0 & \dots \\ \vdots & \vdots & \vdots & \dots \end{pmatrix} \begin{pmatrix} h_1^k(\mathbf{R}) \\ h_2^k(\mathbf{R}) \\ h_3^k(\mathbf{R}) \\ \vdots \end{pmatrix} = \begin{pmatrix} 0 \\ 0 \\ 0 \\ \vdots \end{pmatrix}$$

$$H_n(\mathbf{R})h_n^k(\mathbf{R}) - \varepsilon_k h_n^k(\mathbf{R}) = 0$$

$$-\frac{\hbar^2}{2\mathbf{M}} \nabla_{\mathbf{R}}^2 h_n^k(\mathbf{R}) + E_n(\mathbf{R})h_n^k(\mathbf{R}) = \varepsilon_k h_n^k(\mathbf{R})$$

$$-\frac{\hbar^2}{2\mathbf{M}} \nabla_{\mathbf{R}}^2 h_n^k(\mathbf{R}) + E_n(\mathbf{R}) h_n^k(\mathbf{R}) = \varepsilon_k h_n^k(\mathbf{R})$$

$$h_n^k(\mathbf{R}) = \sum_j c_{nj}^k \chi_{nj}(\mathbf{R})$$

Returning to χ

$$-\frac{\hbar^2}{2\mathbf{M}} \nabla_{\mathbf{R}}^2 \left(\sum_j c_{nj}^k \chi_{nj}(\mathbf{R}) \right) + E_n(\mathbf{R}) \left(\sum_j c_{nj}^k \chi_{nj}(\mathbf{R}) \right) = \varepsilon_k \left(\sum_j c_{nj}^k \chi_{nj}(\mathbf{R}) \right)$$

$$-\frac{\hbar^2}{2\mathbf{M}} \nabla_{\mathbf{R}}^2 c_{nj}^k \chi_{nj}(\mathbf{R}) + E_n(\mathbf{R}) c_{nj}^k \chi_{nj}(\mathbf{R}) = \varepsilon_k c_{nj}^k \chi_{nj}(\mathbf{R})$$

$$-\frac{\hbar^2}{2\mathbf{M}} \nabla_{\mathbf{R}}^2 \chi_{nj}(\mathbf{R}) + E_n(\mathbf{R}) \chi_{nj}(\mathbf{R}) = \varepsilon_{nj} \chi_{nj}(\mathbf{R})$$

Time-independent BO adiabatic formulation

Nuclear Schrödinger equation

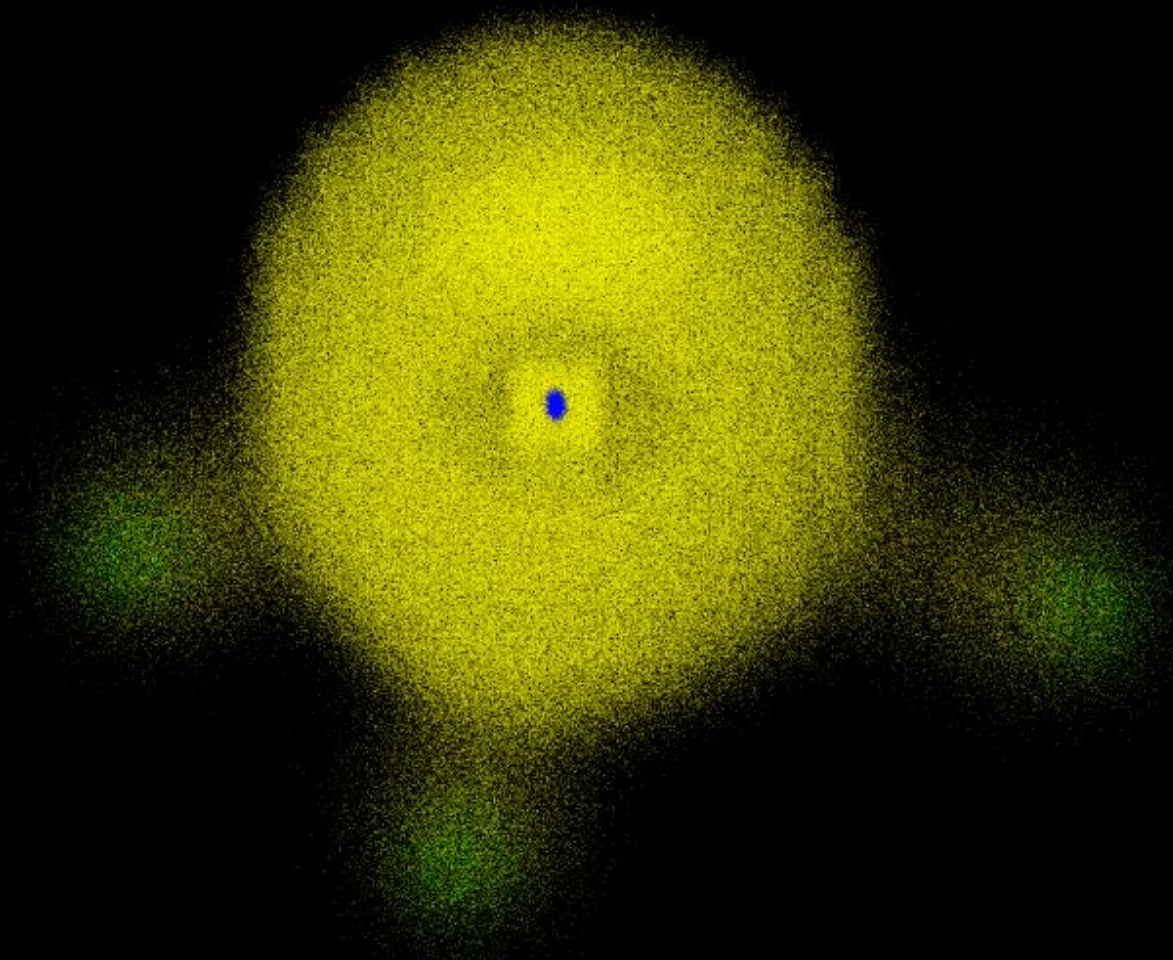
$$\left(T_{nuc}(\mathbf{R}) + E_n(\mathbf{R})\right) \chi_{nj}(\mathbf{R}) = \varepsilon_{nj} \chi_{nj}(\mathbf{R})$$

Electronic Schrödinger equation

$$\left(T_{elec}(\mathbf{r}) + V(\mathbf{r}, \mathbf{R})\right) \varphi_n(\mathbf{r}; \mathbf{R}) = E_n(\mathbf{R}) \varphi_n(\mathbf{r}; \mathbf{R})$$

BO molecular wave function

$$\Psi_{nj}^{BO}(\mathbf{R}, \mathbf{r}) = \varphi_n(\mathbf{r}; \mathbf{R}) \chi_{nj}(\mathbf{R})$$



To know more:

Wave functions and the Schrödinger equation

- McQuarrie; Simon, *Physical chemistry*. **1997**. Ch 3

BO approximation

A math reminder:

- Szabo; Ostlund, *Modern quantum chemistry*. **1989**. Ch 1

The BO approximation

- Eric J Heller, *The semiclassical way*, **2018**. Ch 16

Available for download at:

amubox.univ-amu.fr/s/xXAiMZrDPb9RMRX

The password will be given in class.